



Source apportionment of methane and nitrous oxide in California's San Joaquin Valley at CalNex 2010 via positive matrix factorization

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Abstract. Sources of methane (CH₄) and nitrous oxide (N₂O) were investigated using measurements from a site in southeast Bakersfield as part of the CalNex (California at the Nexus of Air Quality and Climate Change) experiment from mid-May to the end of June 2010. Typical daily minimum mixing ratios of CH₄ and N₂O were higher than daily minima that were simultaneously observed at a mid-oceanic background station (NOAA, Mauna Loa) by approximately 70 ppb and 0.5 ppb, respectively. Substantial enhancements of CH₄ and N₂O (hourly averages >500 and >7 ppb, respectively) were routinely observed, suggesting the presence of large regional sources. Collocated measurements of carbon monoxide (CO) and a range of volatile organic compounds (VOCs) (e.g., straight-chain and branched alkanes, cycloalkanes, chlorinated alkanes, aromatics, alcohols, isoprene, terpenes and ketones) were used with a positive matrix factorization (PMF) source apportionment method to estimate the contribution of regional sources to observed enhancements of CH₄ and N₂O.

The PMF technique provided a “top-down” deconstruction of ambient gas-phase observations into broad source categories, yielding a seven-factor solution. We identified these emission source factors as follows: evaporative and fugitive; motor vehicles; livestock and dairy; agricultural and soil management; daytime light and temperature driven; non-vehicular urban; and nighttime terpene biogenics and anthropogenics. The dairy and livestock factor accounted for the majority of the CH₄ (70–90 %) enhancements during the du-

ration of experiments. The dairy and livestock factor was also a principal contributor to the daily enhancements of N₂O (60–70 %). Agriculture and soil management accounted for ~20–25 % of N₂O enhancements over a 24 h cycle, which is not surprising given that organic and synthetic fertilizers are known to be a major source of N₂O. The N₂O attribution to the agriculture and soil management factor had a high uncertainty in the conducted bootstrapping analysis. This is most likely due to an asynchronous pattern of soil-mediated N₂O emissions from fertilizer usage and collocated biogenic emissions from crops from the surrounding agricultural operations that is difficult to apportion statistically when using PMF. The evaporative/fugitive source profile, which resembled a mix of petroleum operation and non-tailpipe evaporative gasoline sources, did not include a PMF resolved-CH₄ contribution that was significant (<2 %) compared to the uncertainty in the livestock-associated CH₄ emissions. The uncertainty of the CH₄ estimates in this source factor, derived from the bootstrapping analysis, is consistent with the ~3 % contribution of fugitive oil and gas emissions to the statewide CH₄ inventory. The vehicle emission source factor broadly matched VOC profiles of on-road exhaust sources. This source factor had no statistically significant detected contribution to the N₂O signals (confidence interval of 3 % of livestock N₂O enhancements) and negligible CH₄ (confidence interval of 4 % of livestock CH₄ enhancements) in the presence of a dominant dairy and livestock factor. The CalNex PMF study provides a measurement-based assessment

of the state CH₄ and N₂O inventories for the southern San Joaquin Valley (SJV). The state inventory attributes ~ 18 % of total N₂O emissions to the transportation sector. Our PMF analysis directly contradicts the state inventory and demonstrates there were no discernible N₂O emissions from the transportation sector in the southern SJV region.

1 Introduction

Methane (CH₄) and nitrous oxide (N₂O) are the two most significant non-CO₂ greenhouse gases (GHGs), contributing about 50 % and 17 % of the total direct non-CO₂ GHG radiative forcing (~ 1 W m⁻²), respectively (Fig. SPM.5; IPCC, 2013). CH₄, with a lifetime of ~ 10 years and global warming potential (GWP) of 34 on a 100-year basis, accounting for climate-carbon feedbacks (Table 8.7, Myhre et al., 2013; Montzka et al., 2011), is emitted by both anthropogenic and natural sources (e.g., wetlands, oceans, termites, etc.). Anthropogenic global CH₄ emissions are due to agricultural activities (enteric fermentation in livestock, manure management and rice; McMillan et al., 2007; Owen and Silver, 2014), the energy sector (oil and gas operations and coal mining), waste management (landfills and wastewater treatment), and biomass burning (some of which is natural) (Smith et al., 2007; NRC, 2010). N₂O has a higher persistence in the atmosphere (lifetime of ~ 120 years) and stronger infrared radiation absorption characteristics than CH₄, giving it a GWP of 298 (Table 8.7, Myhre et al., 2013; Montzka et al., 2011). Agriculture is the biggest source of anthropogenic N₂O emissions since the use of synthetic fertilizers and manure leads to microbial N₂O emissions from soil (Crutzen et al., 2008; Galloway et al., 2008). Management of livestock and animal waste is another important biological source of N₂O, while industrial processes including fossil fuel combustion have been estimated to account for ~ 10 % of total global anthropogenic N₂O emissions (Denman et al., 2007).

In 2006, the state of California adopted Assembly Bill 32 (AB32) into a law known as the Global Warming Solutions Act, which committed the state to capping and reducing anthropogenic GHG emissions to 1990 levels by 2020. A statewide GHG emission inventory (CARB, 2013) maintained by the Air Resources Board of California (CARB) is used to report, verify and regulate emissions from GHG sources. In 2011, CH₄ accounted for 32.5 million metric tons (MMT) CO₂-eq, representing 6.2 % of the statewide GHG emissions, while N₂O emissions totaled 13.4 MMT CO₂-eq, representing about 3 % of the GHG emission inventory (Fig. 1). CARB's accurate knowledge of GHG sources and statewide emissions is a key component to the success of any climate change mitigation strategy under AB32. CARB's GHG inventory is a "bottom-up" summation of emissions derived from emission factors and activity data. The bottom-

up approach is reasonably accurate for estimation and verification of emissions from mobile and point sources (vehicle tailpipes, power plant stacks, etc.) where the input variables are well understood and well quantified. The main anthropogenic sources of CH₄ in the CARB inventory are ruminant livestock and manure management, landfills, wastewater treatment, fugitive and process losses from oil and gas production and transmission, and rice cultivation, while the major N₂O sources are agricultural soil management, livestock manure management and vehicle fuel combustion (CARB, 2013). The emission factors for many of these sources have large uncertainties as they are biological in nature and their production and release mechanisms are inadequately understood, thus making these sources unsuitable for direct measurements (e.g., emissions of N₂O from farmlands). Many of these sources (e.g., CH₄ from landfills) are susceptible to spatial heterogeneity and seasonal variability. Unfortunately, a more detailed understanding of source characteristics is made difficult because CH₄ and N₂O are often emitted from a mix of point and area sources within the same source facility (e.g., dairies in the agricultural sector), making bottom-up estimation uncertain. There is a lack of direct measurement data or "top-down" measurement-based approaches to independently validate seasonal trends and inventory estimates of CH₄ and N₂O in California's Central Valley, which has a mix of several agricultural sources and oil and gas operations, both of which are known major sources of GHGs.

In the recent past, regional emission estimates derived from measurements from a tall tower at Walnut Grove in central California coupled with inverse dispersion techniques (Fischer et al., 2009) reported underestimation of CH₄ and N₂O emissions especially in the Central Valley. Comparison of regional surface footprints determined from the combination of Weather Research and Forecasting model with Stochastic Time-Inverted Lagrangian Transport algorithm (WRF-STILT) between October and December 2007 indicates that posterior CH₄ emissions are higher than California-specific inventory estimates by 37 ± 21 % (Zhao et al., 2009). Predicted livestock CH₄ emissions are 63 ± 22 % higher than a priori estimates. A study over a longer period (December 2007 to November 2008) at the same tower (Jeong et al., 2012a) generated posterior CH₄ estimates that were 55–84 % larger than California-specific prior emissions for a region within 150 km from the tower. For N₂O, inverse estimates for the same sub-regions (using either EDGAR32 or EDGAR42 a priori maps) were about twice as much as a priori EDGAR inventories (Jeong et al., 2012b). Recent studies have incorporated WRF-STILT inverse analysis on airborne observations across California (Santoni et al., 2012). The authors conclude that CARB's CH₄ budget is being underestimated by a factor of 1.64, with aircraft-derived emissions from cattle and manure management, landfills, rice, and natural gas infrastructure being around 75, 22, 460 and 430 % more than CARB's current estimates for these categories, respectively. Statistical source

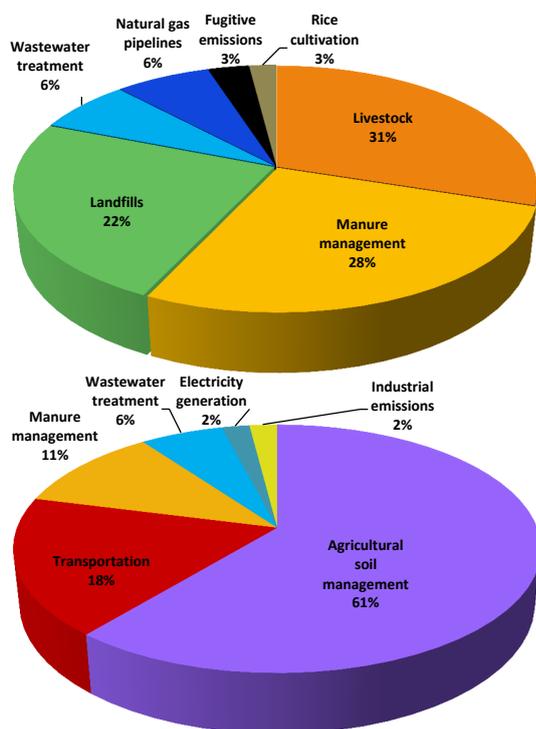


Figure 1. 2011 California emission inventory for (top) methane (CH₄) – 32.5 million metric tons CO₂-eq at GWP = 25 – and (bottom) nitrous oxide (N₂O) – 13.4 million metric tons CO₂-eq at GWP = 298. (Source: CARB GHG Inventory Tool, August 2013)

footprints of CH₄ emissions generated using FLEXPART (FLEXible PARTicle dispersion model)-WRF modeling and CalNex (California at the Nexus of Air Quality and Climate Change) Bakersfield CH₄ concentration data are consistent with locations of dairies in the region (Gentner et al., 2014a). The authors conclude that the majority of CH₄ emissions in the region originate from dairy operations. Scaled-up CH₄ rice cultivation estimates derived from aircraft CH₄/CO₂ flux ratio observations over rice paddies in the Sacramento Valley during the growing season when emissions are at their strongest (Peischl et al., 2012) are around 3 times larger than inventory estimates. CH₄ budgets derived for the Los Angeles (LA) Basin from aircraft observations (Peischl et al., 2013) and studies involving comparison with CO enhancements and inventory at Mt. Wilson (Hsu et al., 2010; Wunch et al., 2009) indicate higher atmospheric CH₄ emissions in the LA Basin than expected from bottom-up accounting.

The recent literature seems to suggest that the CARB bottom-up inventory is underestimating CH₄ and N₂O sources, especially from the livestock sector and perhaps from the oil and gas industry as well. Source apportionment studies of non-CO₂ GHGs over the Central Valley can provide critical information about under-inventoried or unknown sources that seek to bridge the gap between bottom-up and top-down methods. GHG emission inventories can po-

tentially be constrained through simultaneous measurements of GHGs and multiple gas species (VOCs) that are tracers of various source categories. This study provides CH₄ and N₂O source attribution during a 6-week study involving a complete suite of continuous GHG and VOC tracer measurements during the CalNex 2010 campaign in Bakersfield, which is located in the southern part of the Central Valley (May to June 2010). The objective of this study is to partition the measured CH₄, N₂O and VOC enhancements into statistically unique combinations using the positive matrix factorization (PMF) apportionment technique. We classify these combinations as plausible source factors based on our prior knowledge of the chemical origin of mutually co-varying groups of VOC tracers found in each statistical combination. We examine the source categorization using observations from source-specific, ground site and airborne measurements and results from other source apportionment studies. We also compare the relative abundance of CH₄ and N₂O enhancements in each source factor with the CARB inventory estimates in order to assess the accuracy of the inventory. We hypothesize that the PMF analysis will be able to parse the atmospheric observations into unique statistical source combinations that, as analysts, we would be able to distinguish from each other on the basis of unique VOC source markers. We should, thus, be able to appropriately attribute the CH₄ and N₂O apportioned to each of these factor profiles to a major source category. We then proceed to answer the scientific question of whether our top-down assessment of the CH₄ and N₂O inventory can improve our understanding of the bottom-up CARB inventory in the region.

2 Experimental setup

2.1 Field site and meteorology

Measurements were conducted from 19 May to 25 June 2010 at the Bakersfield CalNex supersite (35.3463° N, 118.9654° W) (Fig. 2) in the southern San Joaquin Valley (SJV) (Ryerson et al., 2013). The SJV represents the southern half of California's Central Valley. It is 60 to 100 km wide and surrounded on three sides by mountains, with the Coastal Ranges to the west, the Sierra Nevada to the east, and the Tehachapi Mountains to the southeast.

The measurement site was located to the southeast of the Bakersfield urban core in Kern County (Fig. 2). The east–west Highway 58 is located ~0.8 km north; the north–south Highway 99 is ~7 km west. The city's main wastewater treatment plant (WWTP) and its settling ponds are located east and south (<2.5 km), respectively. Numerous dairy and livestock operations are located south-southwest at 10 km distance and onwards. The metropolitan region has three major oil refineries located within 10 km (two to the northwest, one to the southeast). The majority of Kern County's high-production active oil fields (>10000

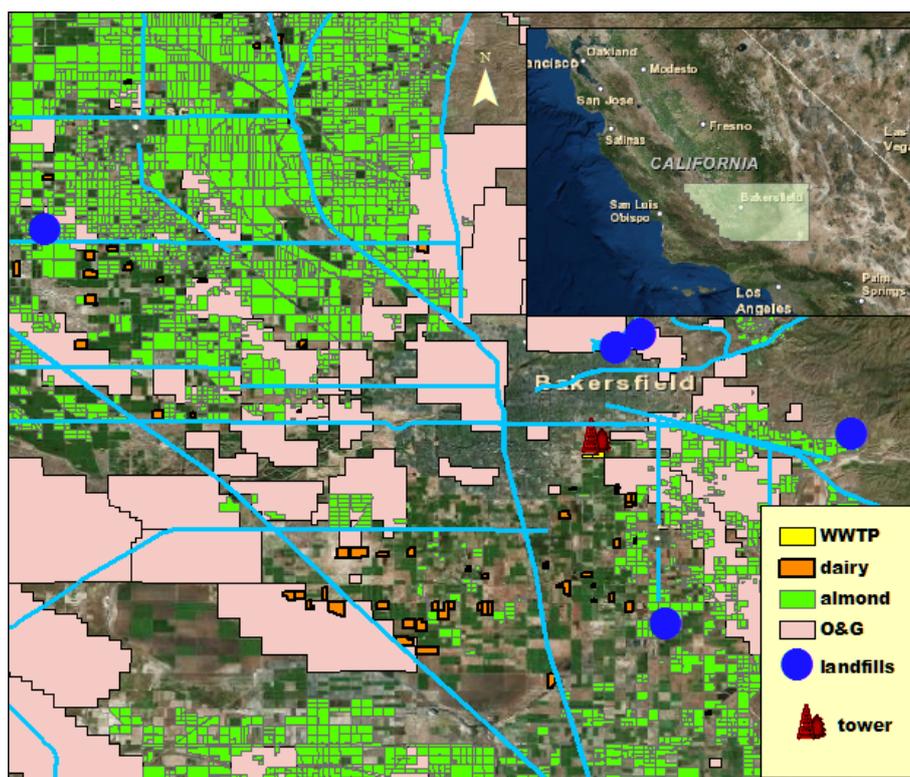


Figure 2. Map of potential sources of methane and nitrous oxide in/around the city of Bakersfield and the surrounding parts of the valley. The inset map is a zoomed-out image of the southern part of San Joaquin Valley (SJV) with location of Kern County superimposed. The light blue lines mark the highways, WWTP stands for wastewater treatment plant, and O&G stands for oil and gas fields. The location of the CalNex experiment site is marked by the “tower” symbol.

barrels (bbl) per day) (CDC, 2013) are located to the west/northwest and are distant ($\sim 40\text{--}100\text{ km}$). Kern River oil field ($\sim 60\,000\text{ bbl day}^{-1}$), one of the largest in the country, and Kern Front ($\sim 11\,000\text{ bbl day}^{-1}$) are located about $10\text{--}15\text{ km}$ to the north. There are several other oil fields dotted within the urban core ($5\text{--}20\text{ km}$) which are less productive ($<2000\text{ bbl day}^{-1}$) or not active ($<100\text{ bbl day}^{-1}$). The whole region is covered with agricultural farmlands, with almonds, grapes, citrus, carrots and pistachios amongst the top commodities by value and acreage (KernAg, 2010).

The meteorology and transport of air masses in the southern SJV is complex and has been addressed previously (Bao et al., 2007; Beaver and Palazoglu, 2009). The wind rose plots (Fig. 3) shown here present a simplified distribution of microscale wind speed and direction at the site for the campaign duration, the latter often being nonlinear over larger spatial scales. The plots depict broad differences in meteorology during daytime and nighttime. A mesoscale representation of the site meteorology during this study period was evaluated through back-trajectory footprints generated from each hourly sample using the FLEXPART Lagrangian transport model with WRF meteorological modeling (Gentner et al., 2014a). The 6 h and 12 h back trajectory footprints are generated at $4 \times 4\text{ km}$ resolution, with simulations originat-

ing from top of the 18 m tall tower. The site experiences persistent up-valley flows from the north and northwest during afternoons and evenings, usually at high wind speeds. The direction and speed of the flow during nights is quite variable (Fig. 3). On some nights, the up-valley flows diminish as nighttime inversion forms a stable layer near the ground, and eventually downslope flows off the nearby mountain ranges bring winds from the east and south during late night and early morning periods. On other nights, fast-moving northwesterly flows extend into middle of the night, leading to unstable conditions through the night. The daytime flows bring plumes from the upwind metropolitan region (Fig. 3) and regional emissions from sources like dairies and farmlands located further upwind. The slow nighttime flows and stagnant conditions cause local source contributions to be more significant than during daytime, including those from nearby petroleum operations and dairies (Gentner et al., 2014a), and agriculture (Gentner et al., 2014b).

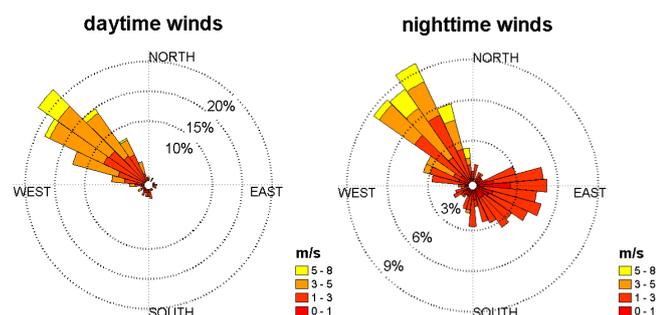


Figure 3. Wind rose plots showing mean wind direction measured at the site during (left) daytime (07:00–16:00 local time, LT) and (right) nighttime (17:00–06:00 LT) during the experiment period in summer 2010. The concentric circles represent the percentage of total observations; each colored pie represents a range of 10° , while the colors denote different wind speed ranges.

3 Methods

3.1 Trace gas measurements and instrumentation

Ambient air was sampled from the top of a tower (18.7 m a.g.l.) through Teflon inlet sampling lines with Teflon filters to remove particulate matter from the gas stream. CH_4 , CO_2 and H_2O were measured using a Los Gatos Research (LGR Inc., Mountain View, CA) Fast Greenhouse Gas Analyzer (FGGA, model 907-0010). N_2O and CO were measured by another LGR analyzer (model 907-0015) with time response of ~ 0.1 to 0.2 Hz. These instruments use off-axis integrated cavity output spectroscopy (ICOS) (O’Keefe, 1998; Paul et al., 2002; Hendriks et al., 2008; Parameswaran et al., 2009). The FGGA instrument internally calculates and automatically applies a water vapor correction to counter the dilution effect of water on a target molecule and calculates CH_4 and CO_2 on a dry (and wet) mole fraction basis. We report dry mole fraction mixing ratios in this study. The FGGA instrument had a 1σ precision of 1 ppb (for CH_4) and 0.15 ppm (for CO_2), while the $\text{N}_2\text{O}/\text{CO}$ instrument had a 1σ precision of 0.3 ppb, over short time periods (< 10 s). Prior to the campaign, the precision of measurements of each instrument used in this study was determined as the 1-sigma standard deviation of a data set over a given length of time measuring a fixed standard (scuba tank) and found to conform to the manufacturer specifications. The instruments were housed at ground level in a thermally insulated, temperature-controlled, 7-foot-wide cargo wagon trailer developed by the instrument manufacturers (Los Gatos Research Inc.). CO was coincidentally measured using another instrument (Teledyne API, USA, Model M300EU2) with a precision of 0.5 % of reading and output as 1 min averages. The mixing ratios from the two collocated CO instruments correlated well ($r \sim 0.99$) and provided a good stability check for the LGR instrumentation. Scaled Teledyne CO data were used to gap-fill the LGR CO data. The coincident

gas-phase VOC measurements were made using a gas chromatograph (GC) with a quadrupole mass selective detector and a flame ionization detector (Gentner et al., 2012).

Hourly calibration checks of the three GHGs and CO were performed using near-ambient-level scuba tank standards through the entire campaign. The scuba tanks were secondary references and were calibrated before and after the experiments using primary standards conforming to the WMO mole fraction scale obtained from the Global Monitoring Division (GMD) at the NOAA Earth System Research Laboratory. The calibration tests confirmed that there was no issue in short-term stability of these species. During data processing, final concentrations were generated from the raw data values using scaling factors obtained from comparison of measured and target concentrations during secondary calibration checks. Diurnal plots of measured species are generated from 1 min averages. PMF analyses in the following sections are based on 30 min averages to match the time resolution of VOC measurements. The meteorological data measured at the top of the tower included relative humidity (RH), temperature (T), and wind speed (WS) and direction (WD).

3.2 Positive matrix factorization (PMF)

Source apportionment techniques like PMF have been used in the past to apportion ambient concentration data sets into mutually co-varying groups of species. PMF is especially suitable for studies where a priori knowledge of the contributing number of sources impacting the measurements, chemical nature of source profiles and relative contribution of each source to the concentration time series of a measured compound is lacking or cannot be assumed. PMF has been applied to ambient particulate matter studies (Lee et al., 1999; Kim et al., 2004); in determining sources of atmospheric organic aerosols (OAs) (Ulbrich et al., 2009; Slowik et al., 2010; Williams et al., 2010); and in gas-phase measurements of VOCs in major metropolitan cities (Brown et al., 2007; Bon et al., 2011). PMF is a receptor-only unmixing model which breaks down a measured data set containing time series of a number of compounds into a mass balance of an arbitrary number of constant source factor profiles (FPs) with varying concentrations over the time of the data set (time series, or TS) (Ulbrich et al., 2009).

In real-world ambient scenarios, emission sources are often not known or well understood. The PMF technique requires no a priori information about the number or composition of factor profiles or time trends of those profiles. The constraint of non-negativity in PMF ensures that all values in the derived factor profiles and their contributions are constrained to be positive, leading to physically meaningful solutions. PMF requires the user to attribute a measure of experimental uncertainty (or weight) to each input measurement. Data point weights allow the level of influence to be related to the level of confidence the analyst has in the measured data (Hopke, 2000). In this way, problematic data such

as outliers, below-detection-limit (BDL) data, or altogether missing data can still be substituted into the model with appropriated weight adjustment (Comero et al., 2009), allowing for a larger input data set, and hence a more robust analysis. PMF results are quantitative; it is possible to obtain chemical composition of sources determined by the model (Comero et al., 2009). PMF is not data-sensitive and can be applied to data sets that are not homogenous and/or require normalization without introducing artifacts.

3.3 Mathematical framework of PMF

The PMF model is described in greater detail elsewhere (Paatero and Tapper, 1994; Paatero, 1997; Comero et al., 2009; Ulbrich et al., 2009), and we will briefly mention some concepts relevant to the understanding of the analysis carried out in this study. The PMF input parameters involve a $m \times n$ data matrix \mathbf{X} with i rows containing mixing ratios at sampling time t_i and j columns containing time series of each tracer j . A corresponding uncertainty matrix \mathbf{S} reports measurement precision (uncertainty) of the signal of each tracer j at every t_i (s_{ij}). The PMF model can then be resolved as

$$\mathbf{X}_{ij} = \sum_p g_{ip} f_{pj} + e_{ij}, \quad (1)$$

where p refers to the number of contributing factors in the solution as determined by the analyst (discussed below), g_{ij} (mass concentration) are elements of a $m \times p$ matrix \mathbf{G} whose columns represent the factor time series, and f_{ij} (mass fraction) are elements of a $p \times n$ matrix \mathbf{F} whose rows represent the factor chemical profiles. e_{ij} are the elements of a $m \times n$ matrix \mathbf{E} containing residuals not fit by the model matrix at each data point.

The PMF algorithm uses a least-squares algorithm to iteratively fit the values of \mathbf{G} and \mathbf{F} by minimizing a “quality of fit” parameter Q (Bon et al., 2011), defined as

$$Q = \sum_{i=1}^m \sum_{j=1}^n (e_{ij}/s_{ij})^2. \quad (2)$$

In this way, PMF minimizes the sum of squares of error-weighted model–measurement deviations. The theoretical value of Q , denoted by Q -expected (Q_{exp}), can be estimated as

$$Q_{\text{exp}} \equiv (m \times n) - p \times (m + n). \quad (3)$$

If all the errors have been estimated within the uncertainty of the data points (i.e., $e_{ij}s_{ij}^{-1} \sim 1$) and the model fits the data perfectly, then Q should be approximately equal to Q_{exp} .

3.4 Data preparation for PMF analysis

For this study, measurements from the FGGA, LGR $\text{N}_2\text{O}/\text{CO}$ analyzer and the GC were combined into a unified data set to create matrices \mathbf{X} and \mathbf{S} . Only VOCs that are

a part of broad chemical composition of nearby sources (like dairies and vehicle emissions) or could potentially serve as source-specific tracers (e.g., iso-octane as a tailpipe emission tracer; isoprene as a biogenic tracer) were included. Isomers were limited (e.g., 2,3-dimethylbutane over 2,2-dimethylbutane), and VOCs with large number of missing values were not included. The input data set represented major chemical families like straight-chain and branched alkanes, cycloalkanes, alkenes, aromatics, alcohols, aldehydes, ketones and chlorinated as well as organosulfur compounds. In spite of best efforts by the authors, it was not possible to quantify the magnitude of observed concentrations of benzene relative to the positive artifacts coming from the Tenax TA adsorbent (previously documented elsewhere). Hence, benzene was not included in the PMF analysis. In all, there were a total of 653 half-hour samples of data collected from 22 May to 25 June. In the days prior to and after this period, there were no N_2O and/or VOC data collected, and hence the PMF analysis is limited to this period. Table 1 lists all the compounds included in the PMF analysis along with a spectrum of observed and background concentrations.

PMF analysis resolves the covariance of mixing ratio enhancements and thus characterizes the chemical composition of emissions from various sources. Hence, for this analysis, only enhancements representing local emissions were included in the data set after subtracting local background concentrations from the original signals. Background concentrations were derived as the minima in the time series (0th percentile) for each of the 50 tracers included in the PMF analysis (CH_4 , N_2O , CO and 46 VOCs). For VOCs, tracers with a minimum value less than 2 times the limit of detection (LOD, in ppt) and a maximum value larger than 100 times the LOD were assumed to have a negligible background (0 ppt) (Table 1). The 99th percentile for each tracer was treated as the effective-maximum mixing ratio and the upper limit of the range for the “normalization” of time series. Enhancements above the 99th percentile are often extreme values. Such outliers, even if true enhancements, represent isolated and short-duration footprints of high-emission events that are difficult for PMF to reconstruct. In order to maintain the robustness of PMF analysis, outliers were selectively down-weighted by increasing their uncertainty in proportion to the uncertainty of other data points (described below). Finally, the enhancements in each time series were “normalized” by dividing every sample by the difference in the 99th percentile and background (the range) as seen in Eq. (4). This process scaled the enhancements in each time series (final data points in \mathbf{X}) within a range of 0 to 1. This allowed for a consistent scheme to represent tracers with vastly different concentrations (e.g., ppm level of CH_4 vs. ppt level of propene) and improve the visual attributes of PMF output plots to follow. Data points denoting zero enhancement (lower limit) were replaced by a very small positive number (i.e., exp^{-5}) to avoid “zeros” in the data matrix \mathbf{X} .

Table 1. PMF data set with total samples (*N*) and mixing ratio range (in ppt).

| Class | Compound | N | 1st percentile | 99th percentile | Background |
|------------------------|---------------------------------|-----|----------------|-----------------|------------|
| GHG | CH ₄ ^{a,c} | 619 | 1855.0 | 3400.8 | 1813.6 |
| | CO ₂ ^{b,c} | 619 | 390.8 | 468.3 | 390.0 |
| | N ₂ O ^{a,d} | 490 | 323.3 | 339.5 | 323.2 |
| Combustion tracer | CO ^{a,d} | 653 | 118.9 | 330.6 | 102.1 |
| Straight chain alkanes | propane | 592 | 580.8 | 30839.0 | 455.5 |
| | <i>n</i> -butane | 587 | 96.4 | 12649.0 | 73.6 |
| | <i>n</i> -pentane | 647 | 93.2 | 3805.4 | 64.4 |
| | <i>n</i> -hexane | 647 | 23.1 | 960.5 | 17.2 |
| | dodecane | 643 | 1.56 | 54.3 | 0 |
| Branched alkanes | isopentane | 646 | 165.4 | 7490.5 | 100.4 |
| | 2,3-dimethylbutane | 650 | 52.5 | 1747.7 | 41.1 |
| | 2,5-dimethylhexane | 651 | 2.37 | 145.8 | 0 |
| | isooctane | 647 | 16.6 | 476.9 | 12.3 |
| | 4-ethylheptane | 651 | 1.45 | 52.6 | 0 |
| | dimethyl undecane | 643 | 0.46 | 24.9 | 0 |
| Cycloalkanes | methylcyclopentane | 647 | 23.3 | 1329.6 | 20.3 |
| | methylcyclohexane | 649 | 8.10 | 813.9 | 0 |
| | ethylcyclohexane | 651 | 1.78 | 169.1 | 0 |
| Alkenes | propene | 592 | 34.7 | 3299.9 | 28.6 |
| | isobutene | 595 | 16.7 | 422.1 | 10.7 |
| Aromatics | toluene | 647 | 48.8 | 1749.5 | 33.1 |
| | ethylbenzene | 647 | 5.83 | 282.0 | 0 |
| | <i>m,p</i> -xylene | 647 | 21.8 | 1127.1 | 21.8 |
| | <i>o</i> -xylene | 647 | 4.31 | 405.0 | 0 |
| | cumene | 640 | 0.55 | 22.8 | 0 |
| | 1-ethyl-3,4-methylbenzene | 651 | 2.22 | 358.6 | 0 |
| | <i>p</i> -cymene | 649 | 0.84 | 93.9 | 0 |
| | indane | 647 | 0.45 | 27.9 | 0 |
| | 1,3-dimethyl-4-ethylbenzene | 635 | 0.46 | 23.9 | 0 |
| | naphthalene | 654 | 0.44 | 19.9 | 0 |
| Unsaturated aldehyde | methacrolein | 573 | 14.2 | 337.0 | 0 |
| Alcohol | methanol | 429 | 2636.81 | 88691.8 | 1085.2 |
| | ethanol | 598 | 1021.93 | 65759.8 | 1021.9 |
| | isopropyl alcohol | 583 | 25.7 | 2001.0 | 25.7 |
| Ketone | acetone | 663 | 142.9 | 3505.8 | 142.9 |
| | methyl ethyl ketone | 605 | 8.55 | 1111.2 | 0 |
| | methyl isobutyl ketone | 629 | 2.03 | 71.9 | 0 |
| Aldehyde | propanal | 636 | 3.68 | 140.8 | 0 |
| | butanal | 589 | 1.72 | 35.1 | 0 |
| Biogenics | isoprene | 651 | 9.70 | 310.0 | 0 |
| | alpha-pinene | 740 | 1.67 | 525.8 | 0 |
| | d-limonene | 641 | 1.10 | 357.1 | 0 |
| | nopinone | 614 | 0.78 | 89.5 | 0 |
| | alpha-thujene | 591 | 0.52 | 23.8 | 0 |
| | camphene | 645 | 0.72 | 100.3 | 0 |
| Chloroalkanes | chloroform | 647 | 34.1 | 209.3 | 31.6 |
| | tetrachloroethylene | 641 | 3.41 | 120.9 | 0 |
| | 1,2-dichloroethane | 640 | 20.6 | 103.8 | 20.6 |
| | 1,2-dichloropropane | 627 | 2.40 | 28.4 | 0 |
| Sulfides | carbon disulfide | 610 | 7.84 | 133.7 | 0 |
| Thiol | ethanethiol | 491 | 4.54 | 685.8 | 0 |

^a parts per billion volume (ppb), ^b Parts per million (ppm), ^c measured using LGR Fast Greenhouse Gas Analyzer, ^d measured using LGR N₂O/CO analyzer.

$$x_{ij} = \frac{(\text{mixing ratio}_{ij} - \text{background}_j)}{(\text{maximum mixing ratio}_j - \text{background}_j)} \quad (4)$$

For the VOCs, guidelines set forth by Williams et al. (2010) were adopted to calculate the uncertainty estimates. An analytical uncertainty (AU) of 10 % was used; a LOD of 1 ppt and a limit of quantification (LOQ) of 2 ppt (Gentner et al., 2012) were used to calculate the total uncertainty for each x_{ij} :

$$s_{ij} \equiv 2 \times \text{LOD}, \text{ if } x_{ij} \leq \text{LOD}, \quad (5a)$$

$$s_{ij} \equiv \text{LOQ}, \text{ if } \text{LOD} < x_{ij} \leq \text{LOQ}, \quad (5b)$$

$$s_{ij} \equiv \left((\text{AU} \times x_{ij})^2 + (\text{LOD})^2 \right)^{0.5}, \text{ if } x_{ij} > \text{LOQ}. \quad (5c)$$

Using this approach, detection limit dictates the errors for low enhancements (near LOD), while errors for larger enhancements of VOCs are tied more to the magnitude of the data value (x_{ij}) itself.

The GHG and CO measurements have high precision and significantly lower detection limits than ambient levels. The relatively low values of GHGs in the uncertainty matrix, compared to VOCs, are substituted with those calculated using a custom approach. The GHG and CO uncertainties are assumed to be proportional to the square root of the data value and an arbitrary scaling factor determined through trial and error in order to produce lower values of QQ_{exp}^{-1} :

$$s_{ij} \equiv A \times (x_{ij})^{0.5}, \text{ where } A = 1 \text{ (for CH}_4\text{)}, 0.25 \text{ (for CO}_2\text{)}, 0.5 \text{ (for CO)}, 0.1 \text{ (for N}_2\text{O)}. \quad (6)$$

This method attributes larger percentage uncertainties to smaller enhancements and hence lesser weight in the final solution and vice versa. This approach leads to an uncertainty matrix that attributes relatively similar percentage errors to both GHGs and VOCs, which should lead to a better fitting of the data through PMF.

Missing values are replaced by the geometric mean of the tracer time series, and their accompanying uncertainties are set at 4 times this geometric mean (Polissar et al., 1998) to decrease their weight in the solution. Based on the a priori treatment of the entire input data (scaling) and the corresponding outputs of the PMF analysis, a weighting approach (for measurements from different instruments) as used in Slowik et al. (2010) is not found to be necessary.

3.5 PMF source analysis

We use the customized software tool (PMF Evaluation Tool v2.04, PET) developed by Ulbrich et al. (2009) in Igor Pro (Wavemetrics Inc., Portland, Oregon) to run PMF, evaluate the outputs and generate statistics. The PET calls upon the PMF2 algorithm (described in detail in Ulbrich et al., 2009)

to solve the bilinear model for a given set of matrices \mathbf{X} and \mathbf{S} for different numbers of factors p and for different values of FPEAK or SEED (defined and described later). The tool also stores the results for each of these combinations in a user friendly interface that allows simultaneous display of FPs and TS of a chosen solution along with residual plots for individual tracers. A detailed explanation of PMF analysis performed in this study is provided in the Supplement (see Section S1–S3). The supplement describes the PMF methodology of how the final number of user-defined factors was chosen (Sect. S1), the outcomes of linear transformations (rotations) of various PMF solutions (Sect. S2) and how uncertainties in the chosen solution were derived (Sect. S3). The standard deviations in the mass fractions of individual tracers in each factor profile and time series of each factor mass is evaluated using a bootstrapping analysis (Norris et al., 2008; Ulbrich et al., 2009) and described in Sect. S3. The uncertainty of a tracer contribution to a source factor is derived from the 1-sigma deviation of the averaged mass fraction of that tracer to that factor from 100 bootstrapping runs. This is the quantity we report and refer to as “uncertainty” throughout Sect. 4. The percentage ranges reported in the Abstract and in Sect. 4 are derived from the relative apportionment of CH_4 and N_2O to different source factors over the 653 half-hourly samples collected during the experiment period. This range represents the mean diurnal range observed, as seen in Fig. 7. This diurnal range combined with bootstrapping-based uncertainty estimates is used to understand better the contribution of each source factor to the observed enhancements of a target GHG and the analyst’s confidence in those estimates.

4 Results and discussion

In Bakersfield, there are a multitude of pollutant sources, ranging from local to regional, from biogenic to anthropogenic, and from primary to secondary. We recognize that PMF analysis is not capable of precise separation of all sources. In PMF analysis, the analyst chooses the number of factor profiles to include in the solution and assigns a source category interpretation for each identified factor. The PMF factors are not unique sources but really statistical combinations of coincident sources. The chemical profile of each factor may contain some contributions from multiple sources that are collocated or may have a similar diurnal pattern of emissions. The cycle of daytime dilution of the boundary layer and nighttime inversion can also result in a covariance that can lead to emissions from unrelated sources being apportioned to a single source factor. Such limitations have been observed previously by Williams et al. (2010) while applying PMF in an urban–industrial setting like Riverside, California. The user must infer the dominant source contributions to these individual factors. Our FP nomenclature is based on the closest explanation of the nature and distribution

of emission sources in the region. The source factor names should be treated with caution, bearing in mind the physical constraints of the solution, and not used to over-explain our interpretation of the region's CH₄ and N₂O inventories.

A seven-factor solution has been chosen to optimally explain the variability of the included trace gases. The factors have been named based on our interpretation of the emission "source" categories they represent, with corresponding colors which remain consistent in the discussion throughout the rest of the paper: evaporative and fugitive (black), dairy and livestock (orange), motor vehicles (red), agricultural + soil management (purple), daytime biogenics + secondary organics (light blue), non-vehicular urban (green), and nighttime anthropogenic + terpene biogenics (navy blue). Figure 4 presents the FP plots of each factor. The sum of the normalized contributions of the 50 species in each source is equal to 1 in the FP plots. Figure 5a–g present the diurnal profiles based on mean hourly concentrations (in normalized units) of each PMF factor, with standard deviations explaining the variability. The interpretation of the individual FPs is discussed below (in Sect. 4.2–4.8). The molar emission factor (EF) of tracers with respect to (w.r.t) one another can be derived for each FP. These EFs can then be compared to those from previous source-specific and apportionment studies (Tables 2–5). The ratio of PMF-derived total CH₄ enhancement to the input measured CH₄ enhancement ranges from 0.90 to 0.95 (mol mol⁻¹) through the whole time series, except for outliers with really high values (> 500 ppb). For N₂O, the ratio is somewhat lower (0.82–0.92 mol mol⁻¹), and this is reflected in the higher PMF-derived uncertainties. The apportionment of some N₂O mass into a statistically weak and time-varying factor is discussed in Sect. 4.5. The general assessment is that PMF analysis is able to reconstruct the majority of the measured enhancements for both CH₄ and N₂O.

4.1 Time trends of measured CH₄, CO₂, CO, and N₂O

The time series of CH₄, CO₂, CO, and N₂O mixing ratios have been plotted in Fig. 6a–d, while the diurnal variations have been plotted in Fig. 6e–h, respectively. The color markers in each plot indicate the median wind direction. The daily minima for the three GHGs and CO occur during the late afternoon period, when daytime heating, mixing and subsequent dilution occur rapidly. The daily minimum values of CH₄ and N₂O were larger than that observed at NOAA's Mauna Loa station at 19.48° N latitude in Hawaii (Dlugokencky et al., 2014) by at least 70 and 0.5 ppb, respectively, for this period. We also compare Bakersfield (at 35.36° N latitude) observations to those from NOAA's Trinidad Head station, which is located on the coast in northern California and is more representative of mid-latitudes at 40.97° N latitude. Although there were no N₂O data collected at Trinidad Head, the CH₄ concentrations observed in discrete flask samples collected every few days during sum-

mer of 2010 (not necessarily a daily low background) were consistently lower than the daily minimum CH₄ concentration curve at Bakersfield by 10–15 ppb. This indicates that there are significant GHG emissions from regional sources around Bakersfield that get added to the already higher local background concentrations, thus keep the local mixing ratio levels quite high. Winds during the highest-temperature period between noon and evening (12:00–20:00 LT) almost always arrive through the urban core in the northwest. Any PMF factor whose dominant source direction is northwest is likely to contain contributions from VOCs emitted from urban sources or regional sources further upwind, or to contain contributions from secondary tracers generated from photochemical processing during the day. The three GHGs show a sharp increase during nighttime, when the inversion layer builds up and traps primary emissions close to the ground. For CO, measured concentrations show two distinct peaks in the diurnal plot (Fig. 6g). The observed early morning peak in the concentration is a combination of decreased dilution and fresh emissions from the morning motor vehicle traffic. The late evening peak in CO concentrations is not coincident with rush hour and is a result of build-up of evening emissions in the boundary layer that gets shallower as the night progresses. Figure 6a indicates CH₄ enhancements of 500 ppb or more on almost every night, with peak mixing ratios exceeding 3000 ppb on several occasions, indicating an active methane source(s) in the region. Figure 6d shows that peak N₂O mixing ratios rise above 330 ppb on almost every night, suggesting large sources in the region. Huge enhancements of CH₄, CO₂ and N₂O (on DOY 157, 164, and 165) (in Fig. 6a, b and d, respectively) may appear well correlated to each other due to regional sources emitting into the inversion layer. However, the shapes of the diurnal cycles differ, indicating different emission distributions, with the early morning maximum in CH₄ occurring before the maxima for CO₂ and N₂O, and the morning maximum for CO occurring slightly later. These differences in timing allow PMF analysis to differentiate their contributions into separate factors.

4.2 Factor 1: evaporative and fugitive emissions

Factor 1 has a chemical signature indicative of evaporative and fugitive losses of VOCs. The FP of this source is dominated by C₃–C₆ straight-chain and branched alkanes and some cycloalkanes (Fig. 4). The average diurnal cycle of factor 1 (Fig. 5a) shows a broad peak during late night and early morning hours, after which the concentrations begin to decrease as the day proceeds, reaching a minimum at sunset before beginning to rise again. This is a strong indication of a source containing primary emissions that build up in the shallow pronounced nighttime inversions of the southern SJV. The subsequent dilution of primary emissions as the mixed layer expands leads to low concentrations during the daytime.

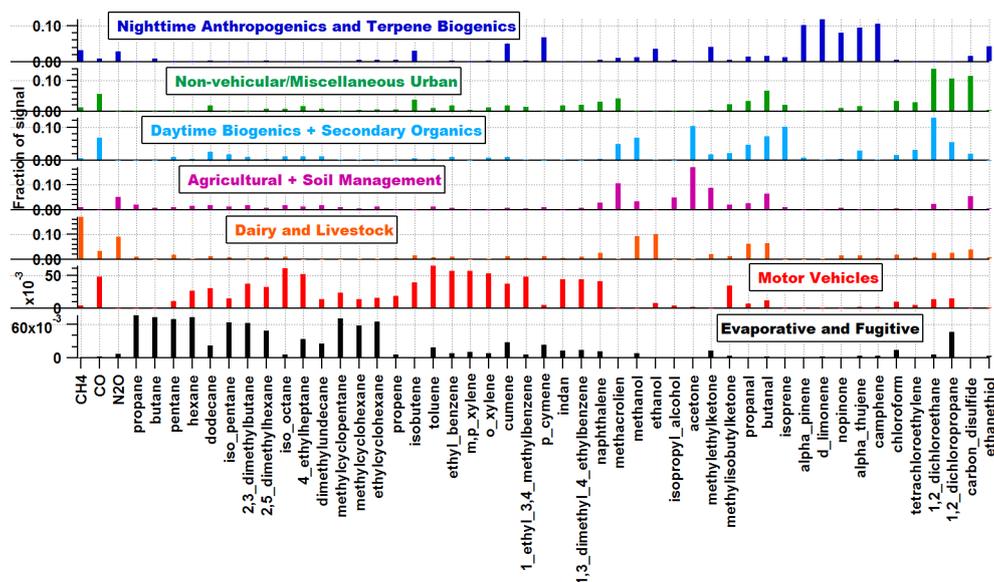


Figure 4. Source profile of the seven factors derived using PMF. The source factors are evaporative and fugitive, motor vehicles, dairy and livestock, agricultural + soil management, daytime biogenics + secondary organics, urban, and nighttime anthropogenics + terpene biogenics. The y axis represents the normalized fraction of mass in each source factor, while the x axis lists all the chemical species included in the PMF analysis.

Table 2. Comparison of ratios of light alkane to propane (gC gC^{-1}) from the PMF fugitive and evaporative factor with those from other PMF studies and oil and gas operations.

| Study | Source | Propane | <i>n</i> -Butane | <i>n</i> -Pentane | <i>n</i> -Hexane | Isopentane |
|---|---------------------------------|---------|------------------|-------------------|------------------|-----------------|
| Bakersfield PMF evaporative and fugitive factor ^a | This study | 1 | 0.52 ± 0.02 | 0.18 ± 0.01 | 0.06 ± 0.003 | 0.33 ± 0.02 |
| Bakersfield petroleum operations source profile ^b | Gentner et al. (2014a) | 1 | 0.53 ± 0.1 | 0.09 ± 0.02 | 0.04 ± 0.01 | 0.08 ± 0.02 |
| Mexico City PMF LPG factor ^c | Bon et al. (2011) | 1 | 0.5 (0.4–0.7) | 0.05 (0.04–0.07) | 0.02 (0.02–0.03) | 0.07 (0.06–0.1) |
| Wattenberg field BAO, Colorado ^d | Gilman et al. (2013) | 1 | 0.75 ± 1.37 | 0.32 ± 0.6 | 0.08 ± 0.13 | 0.28 ± 0.52 |
| Wattenberg field BAO, Colorado ^e | Petron et al. (2012) | 1 | 0.58–0.65 | 0.22–0.31 | NA | 0.22–0.31 |
| PMF natural gas and evaporation factor, Houston Ship Channel ^g | Leuchner and Rappenglück (2010) | 1 | 0.33 | 0.27 | 0.12 | 0.37 |
| PMF natural gas factor, Houston Ship Channel ^h | Buzcu and Fraser (2006) | 1 | 0.67 ± 0.16 | 0.07 ± 0.18 | NA | NA |

^a Uncertainties calculated from propagation of errors (standard deviations) over FPEAK range of -1.6 to $+0.4$. ^b Ratios calculated from Table 4, Gentner et al. (2014a); uncertainties defined as $\pm 20\%$ to account for variability in oil well data. ^c Uncertainties calculated from propagation of uncertainties over FPEAK range of -3 to 3 . ^d Emission ratios derived from multivariate regression analysis; error bars derived from propagation of uncertainty using mean and standard deviation of samples. ^e Range over five regressions conducted over data collected in different seasons and from mobile lab samples. ^f Ratios derived from mean and standard deviations, with propagation of uncertainty. ^g Estimated from Fig. 2, Leuchner and Rappenglück (2010). ^h Estimated from Fig. 2, Buzcu and Fraser (2006).

Most of the propane, *n*-butane and pentane signal is apportioned to this factor, but not the typical vehicle emission tracers like isooctane or CO or any of the alkenes or aromatics. The absence of these tracers in the FP suggests this

factor is not related to vehicular exhaust and is a combination of non-tailpipe emissions and fugitive losses from petroleum operations. None of the CH₄ signal at the SJV site is apportioned to this factor, but almost all of the small straight-chain

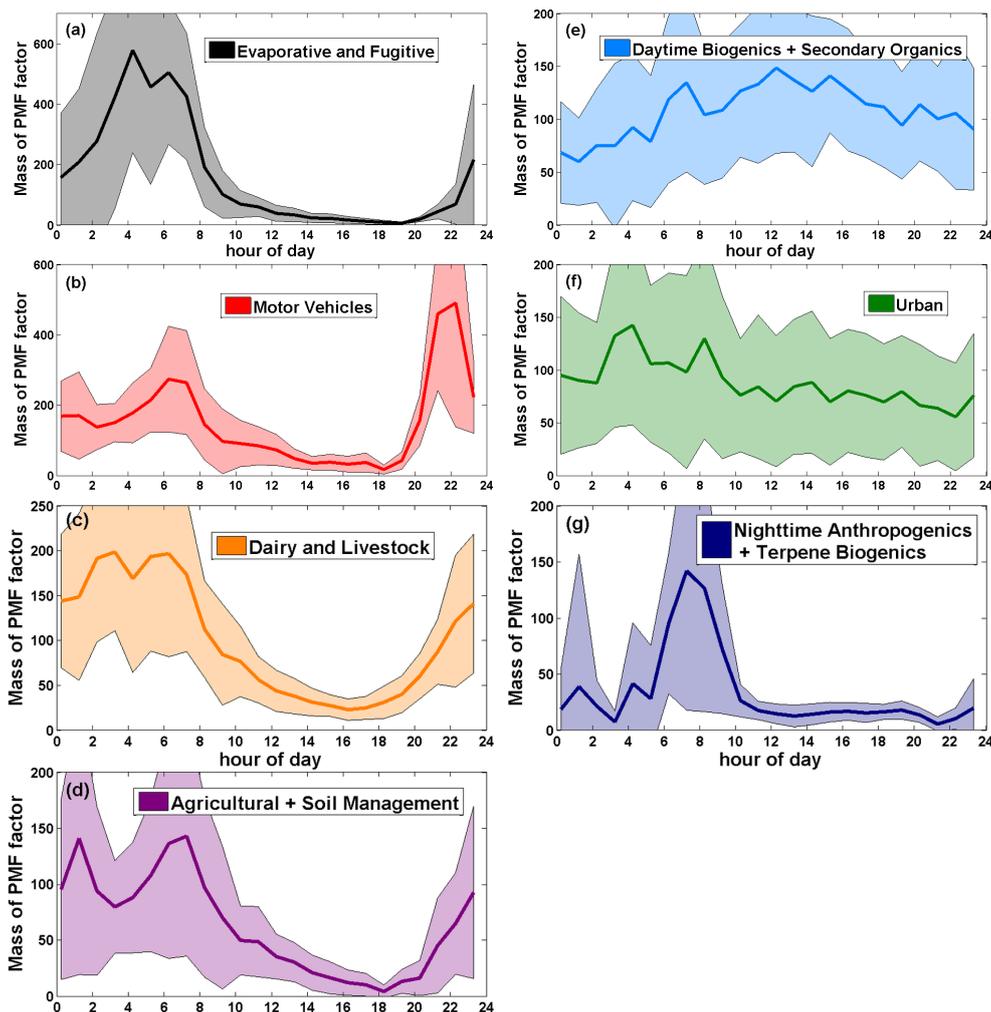


Figure 5. Mean hourly diurnal plots of PMF source factor concentration enhancements for (a) evaporative and fugitive, (b) motor vehicles, (c) dairy and livestock, (d) agricultural + soil management, (e) daytime biogenics and secondary organics, (f) non-vehicular/miscellaneous urban and (g) nighttime anthropogenics + terpene biogenics. The y axis represents the sum of normalized mass concentrations from all tracers contributing to the factor. The x axis is hour of day (local time). The solid lines represent the mean, and the shaded area represents the standard deviation (variability) at each hour.

alkanes exclusively apportion to this factor. This is in agreement with Gentner et al. (2014a), where the authors show that VOC emissions from petroleum operations are due to fugitive losses of associated gas from condensate tanks following separation from CH_4 . Table 2 compares EFs derived from this PMF study for the non-tailpipe (evaporative) and fugitive petroleum operation source factor with those from the Gentner et al. (2014a) study done on the same CalNex data set using an independent source receptor model with chemical mass balancing and effective variance weighting method, and also to reports of fugitive emissions from the oil and natural gas sources (Pétron et al., 2012; Gilman et al., 2013) and similar factors produced by other PMF studies (Buzcu and Fraser, 2006; Leuchner and Rappenglück, 2010; Bon et al., 2011). Good agreement of factor 1 VOC EFs with those from the mentioned studies confirms petroleum oper-

ations in Kern County as the major source contributing to this factor. The PMF apportionment indicates that this source factor does not contribute to CH_4 enhancements observed at the SJV site (Fig. 7a) and thus that most of the “associated” CH_4 is likely separated from the condensate prior to emission. As mentioned before, a tiny fraction ($\sim 5\%$; Sect. 4) of the total input CH_4 enhancement is not resolved into source-apportioned contributions. There could be a minor contribution to CH_4 signal from this source, which is unresolved within the framework of uncertainties in the PMF analysis.

4.3 Factor 2: motor vehicle emissions

Factor 2 has a chemical signature consistent with the tailpipe exhausts of gasoline and diesel motor vehicles. This source factor includes the combustion tracer CO and other vehic-

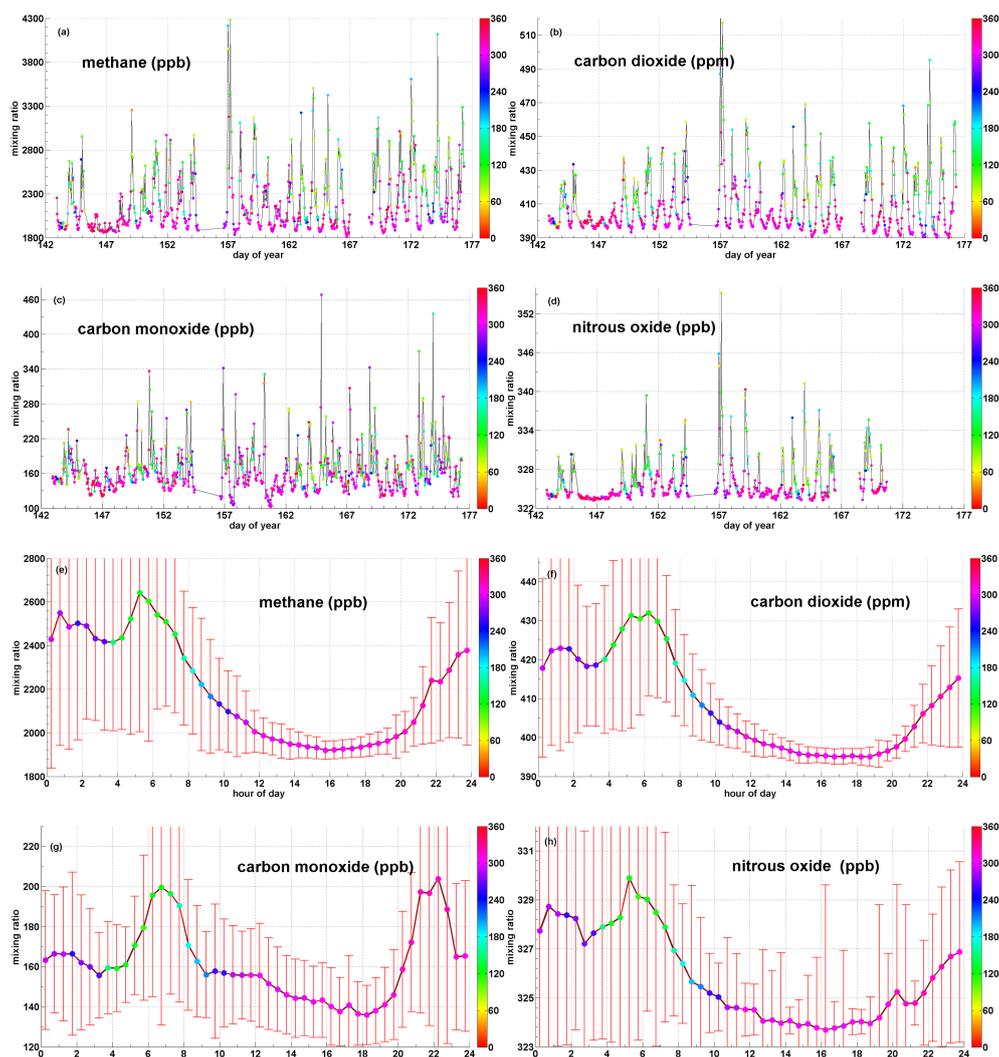


Figure 6. Time series of (a) CH₄, (b) CO₂, (c) CO, and (d) N₂O obtained from 30 min averages from 15 May to 30 June 2010. The color bar indicates the average wind direction during each 30 min period. Mixing ratios plotted as average diurnal cycles for (e) CH₄, (f) CO₂, (g) CO and (h) N₂O along with wind direction. The curve and the red whiskers represent the mean and the standard deviations about the mean, respectively.

ular emission tracers, such as isooctane (Fig. 4). Alkenes are a product of incomplete fuel combustion in motor vehicles, and almost all of the propene and a significant portion of the isobutene signal are attributed to this source factor. The diurnal variation of factor 2 shows two distinctive peaks (Fig. 5b). The first peak occurs in the morning between 06:00 and 07:00 LT and is influenced by morning rush hour traffic, with suppressed mixing allowing vehicle emissions to build up. As the day proceeds, accelerated mixing and dilution (and perhaps chemical processing of reactive VOCs) reduce the enhancements to a minimum by late afternoon. The evening peak mainly occurs as the dilution process slows down after sunset and emissions build up. The increased motor vehicle traffic in the evening adds more emissions to the shrinking boundary layer. This build-up reaches a

peak around 22:00 LT. The occasional high-wind events from the northwest (unstable conditions) and fewer vehicles operating on the roads during late nighttime hours contribute to the relatively lower levels of enhancements as compared to the peaks on either side of this nighttime period.

Table 3 compares selective PMF-derived EFs from vehicle emissions with the measured gasoline composition collected during CalNex in Bakersfield (Gentner et al., 2012), analysis of gasoline samples from Riverside in the Los Angeles Basin (Gentner et al., 2009) and ambient VOC emission ratios measured during CalNex at the Pasadena supersite (Borbon et al., 2013). Although the two Bakersfield studies employ different source apportionment techniques (and so do the studies conducted in the Los Angeles Basin), we observe a broad agreement of relative emission rates of vehicular emission

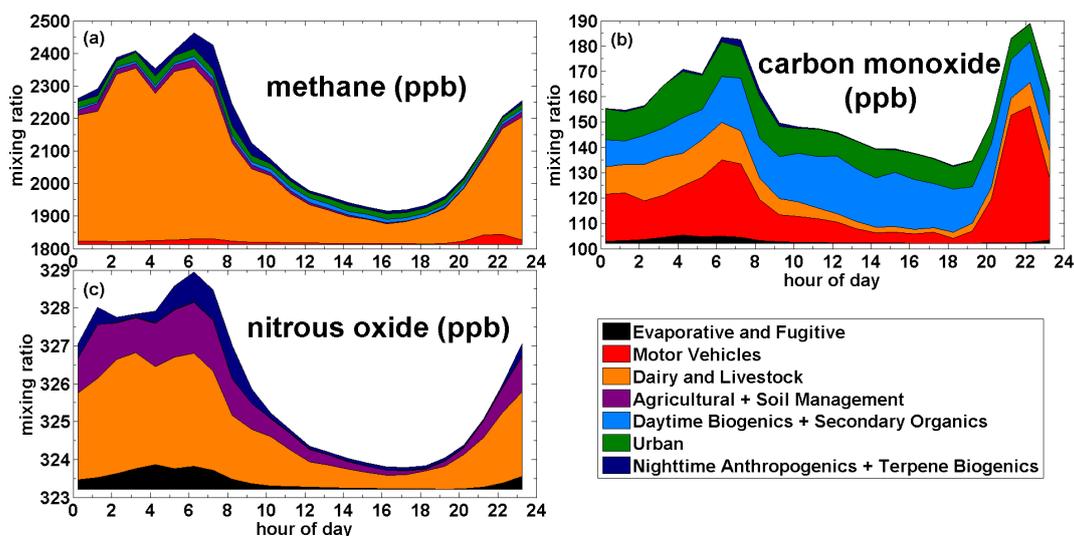


Figure 7. Diurnal plot of PMF-derived (a) CH_4 , (b) CO , and (c) N_2O concentrations sorted by PMF source category. The legend on the bottom right shows the names of the PMF source factor which each color represents. The PMF-derived enhancements from each source have been added to the background concentrations.

tracers. This agreement validates our assertion that factor 2 represents a broad suite of vehicular tailpipe emissions.

The PMF-derived CH_4/CO EF in factor 2 is $0.58 \text{ (mol mol}^{-1}\text{)}$ and is significantly higher than the range of $0.03\text{--}0.08 \text{ (mol mol}^{-1}\text{)}$ calculated from results of a vehicle dynamometer study of 30 different cars and trucks (Nam et al., 2004) and an EF of $0.014 \text{ (mol mol}^{-1}\text{)}$ calculated for SJV district during summer of 2010 using Emission Factors (EMFAC), which is CARB's model for estimating emissions from on-road vehicles operating in California (EMFAC, 2011). While it is certainly a possibility that the current in-use CH_4 emission factor in the inventory may be an underestimation, it seems more logical that the relatively high proportion of CH_4 signal in the vehicle source factor profile is due to contributions from coincident urban sources (e.g., natural gas leaks) mixed into the vehicle gasoline exhausts, resulting in a “mixing” phenomena as discussed in the Supplement. In spite of the non-negligible proportion of CH_4 in the factor 2 source profile, the contribution of the factor to CH_4 enhancements (Fig. 7a) at Bakersfield is negligible relative to the dairy and livestock factor.

The state GHG inventory attributes about 18 % of the 2010 statewide N_2O emissions to the on-road transportation sector (CARB 2012). Our PMF analysis shows essentially a negligible enhancement of N_2O associated with the vehicle emission factor 2 with a PMF-derived $\text{N}_2\text{O}/\text{CO}$ EF of $0.00015 \text{ (mol mol}^{-1}\text{)}$. The EMFAC-generated $\text{N}_2\text{O}/\text{CO}$ EF in the SJV during summer of 2010 is more than 20 times higher at $0.0034 \text{ (mol mol}^{-1}\text{)}$. The PMF-derived “vehicle emission” contribution to N_2O is in stark contrast to the inventory and is an important outcome suggesting a significant error in EFs used to derive the statewide inventory for N_2O .

4.4 Factor 3: dairy and livestock emissions

Factor 3 has a chemical signature indicative of emissions from dairy operations. This source factor is the largest contributor to CH_4 enhancements (Fig. 7a) and a significant portion of the N_2O signal (Fig. 7c). The FP also has major contributions from methanol (MeOH) and ethanol (EtOH), with minor contributions from aldehydes and ketones (Fig. 4). A separate PMF analysis with a broader set of VOC measurements at the same site showed that most of the acetic acid (CH_3COOH) and some formaldehyde (HCHO) signal attributed to this factor as well, personal communication, 2014). All the abovementioned VOCs are emitted in significant quantities from dairy operations and cattle feedlots (Filipy et al., 2006; Shaw et al., 2007; Ngwabie et al., 2008; Chung et al., 2010). About 70–90 % of the diurnal CH_4 signal is attributed to this factor (Fig. 7a) depending on the time of day. The CH_4 dairy and livestock mass fraction which is used to calculate this diurnal range has an uncertainty of 29 %, derived using the bootstrapping method. This source factor contributes about 60–70 % of the total N_2O daily enhancements as seen in Fig. 7c. The bootstrapping uncertainty in the N_2O dairy and livestock mass fraction is 33 %.

Comparing the factor 3 profile to dairy source profiles from various studies is challenging. A dairy is, in essence, a collection of area sources with distinct emission pathways and chemical characteristics. Hence, a lot of dairy studies do not look at facility-wide emissions, instead focusing on specific area sources within the facility. In contrast, PMF captures the covariance of CH_4 , N_2O , and VOCs emitted from the ensemble source as downwind plumes from dairies arrive at the site. Table 4 compares the PMF-derived EFs of CH_4 w.r.t MeOH and EtOH with those from other studies. Previ-

Table 3. Comparison of ratios of hydrocarbon to toluene (gC gC^{-1}) from the PMF vehicle emission factor with similar ratios from other California specific studies.

| Study | Bakersfield PMF vehicle emissions factor ^a | Bakersfield gasoline source profile ^{b,c} | Riverside liquid gasoline profile ^e | CalNex Los Angeles ambient emission ratios ^g |
|--------------------|---|--|--|---|
| Source | This study | Gentner et al. (2012) | Gentner et al. (2009) | Borbon et al. (2013) |
| CH ₄ | 8.1 ± 2.1 | NA | NA | NA |
| CO | 14.0 ± 0.4 | NA | NA | 45 |
| Toluene | 1 | 1 | 1 | 1 |
| Isopentane | 0.69 ± 0.01 | 0.77 ± 0.04 | 0.64–0.84 | 1.95 |
| Isooctane | 0.29 ± 0.03 | 0.34 ± 0.02 | 0.64–0.80 | NA |
| <i>n</i> -Dodecane | 0.03 ± 0.001 | (0.02 ± 0.007) ^d | NA | NA |
| Methylcyclopentane | 0.24 ± 0.01 | 0.32 ± 0.02 | NA | NA |
| Ethyl benzene | 0.17 ± 0.01 | 0.14 ± 0.01 | NA | 0.2 |
| <i>m/p</i> -Xylene | 0.65 ± 0.01 | 0.65 ± 0.03 | (0.45–0.52) ^f | 0.64 |
| <i>o</i> -Xylene | 0.22 ± 0.01 | 0.23 ± 0.01 | NA | 0.24 |

^a Errors are standard deviation of 12 unique PMF solutions over FPEAK range of -1.6 to $+0.4$; see Sect. S2. ^b Derived from liquid gasoline fuel speciation profile (Table S9; Gentner et al., 2012). ^c Errors bars derived from propagation of uncertainties. ^d Derived by combining diesel fuel and gasoline speciation profile (Tables S9 and S10; Gentner et al., 2012) and gasoline and diesel fuel sale data in Kern County (Table S1, Gentner et al., 2012). ^e Summer data. ^f Only *m*-xylene. ^g Derived from linear regression fit slope of scatterplot from CalNex Pasadena supersite samples.

ously, cow chamber experiments (Shaw et al., 2007; Sun et al., 2008) have measured emissions from ruminants and their fresh manure; emissions have also been studied in a German cowshed (Ngwabie et al., 2008), and EFs have been derived from SJV dairy plumes sampled from aircraft (Gentner et al., 2014a; Guha et al., 2015). Since enteric fermentation and waste manure are the predominant CH₄ source in dairies, CH₄ emission rates calculated by Shaw et al. (2007) are representative of a whole facility. However, their MeOH/CH₄ ratios are lower than those determined by PMF and aircraft studies. Animal feed and silage are the dominant sources of many VOCs, including MeOH and EtOH (Alanis et al., 2010; Howard et al., 2010), and the ratios in Shaw et al. (2007) do not reflect these emissions. In Ngwabie et al. (2008), experiments were performed in cold winter conditions (-2 to 8 °C), when temperature-dependent VOC emissions from silage and feed are at a minimum. The authors comment that MeOH emissions from California dairies are likely higher, as the alfalfa-based feed is a big source of MeOH owing to its high pectin content (Galbally and Kirstine, 2002). These observations explain why MeOH/CH₄ ratios in these studies are lower than PMF-derived ratios. The PMF range for the EtOH/CH₄ EF for factor 3 agrees with the slope derived from ground-site data (Gentner et al., 2014a) and is similar to, but somewhat larger than, the German dairy study (Ngwabie et al., 2008). Miller and Varel (2001) and Filipy et al. (2006) did not measure CH₄ emission rates, so a direct derivation of EF w.r.t CH₄ is not possible. These studies, however, reported EtOH emission rates (from dairies and feedlots in the United States) which are used to derive EFs w.r.t CH₄ using an averaged CH₄ emission rate from Shaw et al. (2007). Using this method, we get EFs that are com-

parable to the PMF-derived EF of CH₄/EtOH (Table 4). Hence, we demonstrate within reasonable terms that the relative fractions of masses in factor 3 are consistent with CH₄ and VOC emissions from dairies.

Enteric fermentation is a part of the normal digestive process of livestock, such as cows, and is a large source of CH₄, while the storage and management of animal manure in lagoons or holding tanks is also a major source of CH₄. According to the state GHG inventory (CARB, 2013), ~ 58 % of the statewide CH₄ emissions results from a combination of these two processes. N₂O is also emitted during the breakdown of nitrogen in livestock manure and urine and accounts for about 10 % of the statewide N₂O emission inventory. Kern County has a big dairy industry, with about 160 000 milk cows representing 10 % of the dairy livestock of the state in 2012 and another 330 000 heads of cattle for beef (KernAg, 2011; CASR, 2013). The dominant contributions to CH₄ and N₂O signal and the general agreement of dairy EFs with PMF EFs from factor 3 indicate that the extensive cattle operations in the county are a big source of these emissions. We do observe that the proportion of regional N₂O enhancements attributed to this sector is a significantly larger proportion of the total N₂O emissions as compared to the state inventory.

4.5 Factor 4: agricultural and soil management emissions

The chemical profile of factor 4 is a mix of emissions from agricultural activities around the site. Factor 4 includes a major portion of the N₂O signal along with a number of VOCs that have crop/plant signatures like methacrolein, methyl ethyl ketone (Jordan et al., 2009; McKinney et al., 2011),

Table 4. Comparison of PMF dairy and livestock emission factors (mmol mol^{-1}) with previous studies.

| Study | Source | Cow/manure type (if applicable) | Methanol/methane EF avg. (range) | Ethanol/methane EF avg. (range) |
|--|-------------------------|---------------------------------|----------------------------------|---------------------------------|
| PMF analysis of regional measurements | This study | | 15–47 | 9–32.2 |
| Environmental chamber with cows and/or manure | Shaw et al. (2008) | Dry | 3.2 (0.6–7.4) | NA |
| Environmental chamber with cows and/or manure | Sun et al. (2008) | Lactating | 1.9 (0.8–3.6) | NA |
| | | Dry | 13.4 (4–25) | 14.4 (11–19) |
| Cowshed with regular dairy operations (winter) | Ngwabie et al. (2008) | Lactating | 19.2 (15–25) | 24.2 (18–32) |
| Cow stall area with regular dairy operations (summer) | Filipy et al. (2006) | | NA | 9.3 (4–16) |
| Manure from cattle feedlot | Miller and Varel (2001) | Fresh (< 24 h) | NA | 14 ^b |
| | | Aged (> 24 h) | | 118 ^b |
| Measured slope of regression (CalNex 2010) | Gentner et al. (2014a) | | 7.4 (7–16) ^c | 18 ^d |
| Sampling of dairy plumes from aircraft (CABERNET 2011) | Guha et al. (2015) | | 9.6 (9–30) ^c | NA |

^a Calculated based on CH_4 emission rate of $4160 \mu\text{g cow}^{-1} \text{s}^{-1}$ for mid-lactating cows (Shaw et al., 2007). ^b Calculated based on CH_4 emission rate of $4160 \mu\text{g cow}^{-1} \text{s}^{-1}$ for mid-lactating cows (Shaw et al., 2007) and ethanol emission rate for fresh and aged manure of 175 and $1223 \mu\text{g cow}^{-1} \text{s}^{-1}$, respectively, derived by Filipy et al. (2006). ^c Slope of regression with range of measured slopes (in parentheses) from sampling of dairy plumes by aircraft. ^d Ground site data; lower limit of slope of non-vehicular ethanol versus methane.

methanol and acetone (Goldstein and Schade, 2000; Hu et al., 2013; Gentner et al., 2014b) (Fig. 4). While many of these oxygenated VOCs have several prominent sources, studies have reported substantial simultaneous emissions from natural vegetation and agricultural crops. At a rural site in the northeastern USA, Jordan et al. (2009) reported high concentrations of oxygenated VOCs and correlations between the diurnal concentrations of acetone, methanol, and methyl ethyl ketone. Kern County is one of the most prolific agricultural counties in California. The four main crops grown (by value as well as acreage) in 2010 were almonds, grapes, citrus and pistachios (KernAg, 2011). Table 5 compares the PMF-derived EFs for acetone/MeOH from factor 4 with ratios of basal emission factors (BEFs) from crop-specific greenhouse and field measurements (Fares et al., 2011, 2012; Gentner et al., 2014b). The good agreement of the ratios confirms that the FP of factor 4 is an aggregate of biogenic VOC emissions from the agricultural sector. Nitrous oxide is emitted when nitrogen is added to soil through use of synthetic fertilizers and animal manure, while crops and plants are responsible for the VOC emissions. Hence this source factor is a combination of collocated sources (soils and crops). The PMF solution to this factor has uncertainties greater than those for other factors (Fig. S4). This is potentially because not all crops emit the same combination of VOCs, nor are all agricultural fields fertilized at the same time. The existence of this statistically weak factor is confirmed by bootstrapping runs (Sect. S3) and numerous PMF trials, all of which produce a distinct factor with N_2O as a dominant contribu-

tor along with certain biogenic VOCs, though often in varying proportions. CO_2 is not included in the PMF analysis reported in the paper, most importantly because negative CO_2 fluxes during daytime can introduce artifacts in PMF analysis and result in erroneous apportionment. But PMF runs involving CO_2 indicate that most of the CO_2 is apportioned to this factor. Plant and soil respiration (especially during the night) is a major source of CO_2 , and the apportionment of CO_2 to factor 4 confirms the nature of this source. The temporal correlation between CO_2 and N_2O is also evident in their average diurnal cycles (Fig. 6f and h), which have a coincident early morning peak. The absence of monoterpenes from the FP of this factor can be explained by their shorter atmospheric lifetimes compared to VOCs like acetone and MeOH and the rapid daytime mixing which dilutes the terpenoid emissions arriving at the site during the day. At night, when atmospheric dilution is low, monoterpene emissions from agriculture are more likely to get apportioned into a separate source factor dominant during nighttime, when temperature-sensitive biogenic emissions of MeOH and acetone can be expected to be a minor constituent in the FP (see Sect. 4.8).

Factor 4 is a significant source of GHGs, contributing about 20–25 % of the total N_2O enhancements in the diurnal cycle (Fig. 7c) but with a relatively large 1σ confidence interval of 70 % in the agriculture and soil N_2O mass fraction. Kern County is one of the premier agricultural counties of California, accounting for USD 4.2 billion (about 18 %) of the total agricultural revenue from fruits, nuts, vegetables and field crops (KernAg, 2011; CASR, 2013) and is also the

Table 5. Comparison of PMF agricultural and soil management emission factor for acetone versus methanol (gC gC^{-1}) with ratios of basal emission factors generated for major crops grown in the Kern County. Errors denote standard deviations computed by propagation of uncertainty.

| Bakersfield PMF agricultural and soil management factor | Almond greenhouse summer 2008 | Table grape greenhouse summer 2008 | Pistachio greenhouse summer 2008 | Navel oranges greenhouse summer 2008* | Valencia oranges greenhouse summer 2008 |
|---|--|---|---|---------------------------------------|---|
| This study 0.58 ± 0.37 | Gentner et al. (2014b) 0.14 ± 0.2 | Gentner et al. (2014b) 0.04 ± 0.02 | Gentner et al. (2014b) 0.5 ± 0.6 | Fares et al. (2011) 0.57 ± 0.1 | Fares et al. (2012) 0.5 ± 0.3 |

* Branch with flowers not removed.

biggest consumer of synthetic fertilizers. Agricultural soil management accounts for about 60% of the statewide N_2O emission inventory (CARB, 2013). Our assessment of diurnal source distribution of N_2O emissions from the agriculture source factor (Fig. 7c) in the presence of another dominating source (dairy and livestock) is consistent with the inventory estimates from the agricultural and soil management sector.

4.6 Factor 5: daytime biogenics and secondary organics

The chemical composition and diurnal profile of factor 5 points to a source whose emissions are either primary biogenic VOCs with temperature-dependent emissions (e.g., isoprene) or products of photochemical oxidation of primary VOCs (e.g., acetone) (Fig. 4). Isoprene is a dominant component of the source FP and is mostly apportioned to factor 5. Figure 5e shows a steady increase in the PMF factor mass concentration during the daytime hours, which hits a peak during afternoons, indicating that this source is dependent on sunlight and temperature. Potential source contributions come from oak forests in the foothills of the western edge of the SJV or scattered isoprene-producing plants in the SJV (note that most crops do not emit significant amounts of isoprene). Factor 5 includes contribution from VOCs that have primary light and temperature-driven sources (e.g., crop emissions), as well as secondary sources in the Central Valley, e.g., acetone (Goldstein and Schade, 2000), methanol (Gentner et al., 2014b) and aldehydes. A similar PMF analysis with a different objective (Goldstein et al., 2015) shows that secondary organics like glyoxal, formaldehyde and formic acid mostly apportion to factor 5. The CO apportioned to this factor could potentially be a product of mobile and/or stationary combustion collocated or up/downwind of the biogenic VOC source. CO can also come from coincident isoprene oxidation (Hudman et al., 2008). This daytime source is not responsible for any of the observed CH_4 and N_2O enhancements.

4.7 Factor 6: non-vehicular/miscellaneous urban emissions

The chemical signature of factor 6 is composed of VOCs associated with an array of applications and processes, including solvents, fumigants, industrial byproducts, etc. The diurnal profile of factor 6 (Fig. 5f) is somewhat different from that of the evaporative and fugitive source (Fig. 5a) and dairies (Fig. 5c) in that, even during the middle of the day when vertical mixing is at its strongest, the enhancements contributing to the factor are substantial. This suggests that the source(s) is in close proximity to the site and hence most likely located within the urban core. The FP has CO as an important component, but the relative absence of fugitive source markers (e.g., light alkanes) and vehicle emission tracers (e.g., isooctane, cycloalkanes, etc.) indicates that the origin of this source factor is potentially non-mobile combustion. Also present in a major proportion is carbon disulfide (CS_2), chlorinated alkanes like 1,2-dichloroethane and 1,2-dichloropropane, isobutene (product of incomplete combustion), and minor contributions from aromatics and aldehydes (Fig. 4). There are a myriad of potential sources that could be contributing to this factor, and we do not have specific tracers or other information to ascribe it to a single source or group of sources. Hence we call factor 6 an “urban emission source”. There is a very minor CH_4 contribution from this factor which results in a tiny and negligible contribution to the PMF source apportionment of CH_4 (Fig. 7a). The source factor does not contribute to the N_2O enhancements.

4.8 Factor 7: nighttime anthropogenic and terpene biogenic emissions

Factor 7 is primarily composed of biogenic compounds belonging to the terpene family and *p*-cymene (Fig. 4). Factor 7 mostly influences the site during late night and early morning hours (Fig. 5g), when nighttime downslope flows usually dominate, bringing winds from the east and south to the site. The entire flow path from the base of the foothills to the site is covered with agricultural crops emitting into a shallow nighttime boundary layer. These crops include grapes, almonds, citrus and pistachios, which are the top four agricultural commodities grown in the county (KernAg, 2011; CASR, 2013),

and these produce considerable monoterpene emissions (Fares et al., 2012; Gentner et al., 2014b). The spatial distribution of terpene compounds from the statistical source footprint derived from FLEXPART back trajectories is consistent with the location of croplands in the southern SJV (Gentner et al., 2014b). Biogenic VOCs emitted from forests in the foothills are likely minor contributors to the downslope flows arriving at the site owing to their lifetime and distance (> 50 km) (Tanner and Zielinska, 1994).

Following the rapid rise in enhancements in the early morning hours, contributions of factor 7 to total signal decrease rapidly when the flow moves to more typical daytime wind directions (Fig. 5g). A nearby source (e.g., the WWTP), which is upwind of the site for only a certain part of the diurnal cycle, is expected to be more directionally constrained, and the emission profile from such a source will look similar to the diurnal profile of factor 7. Among source factors which contain a non-negligible fractional contribution of both CH₄ and N₂O (i.e., dairies, agriculture and soil management, and factor 7), the PMF-derived CH₄/N₂O EF of 42 ± 20 (gC gN⁻¹) from factor 7 is most similar to the bottom-up inventory EF of 56 (gC gN⁻¹) for wastewater treatment in Kern County (KernGHG, 2012). Given the proximity of the WWTP and previous observations of GHGs from them, it is possible that there is a minor but noticeable contribution (~ 5 %) to CH₄ and N₂O enhancements from this nighttime source (Fig. 7a and c).

5 Implications

This study demonstrates the potential of the PMF technique to apportion atmospheric gas-phase observations of CH₄ and N₂O into source categories using a broad array of tracers. PMF is not commonly employed to perform source attribution of these GHGs because studies generally lack simultaneous measurements of specific source markers. Applying this statistical technique on a GHG–VOC unified data set, well represented by a broad suite of VOC classes, allows a set of compounds acting as source markers to be partitioned into separate profiles, leading to easier identification of their sources.

We provide clear analysis that dairy and livestock operations are the largest sources of emissions in the Bakersfield region, accounting for the majority of the CH₄ (70–90 %) and N₂O (60–70 %) emissions. As per the CARB inventory (Fig. 1), dairy operations are the dominant source of non-CO₂ GHGs in the state, and our analysis agrees with that broad trend. However, in the recent past, a number of top-down CH₄ and N₂O emission studies in the Central Valley have reported underestimation of the non-CO₂ GHG inventory (Zhao et al., 2009; Santoni et al., 2012; Jeong et al., 2012a, b; Miller et al., 2013). These studies attribute the majority of this underestimation to the dairy sector. Our results emphasize the significance of this sector in the SJV, although

we did not derive total emission estimates to compare directly with the inventory.

The contribution of fugitive emissions from the oil and gas industry in Bakersfield to CH₄ emissions is found to be negligible especially in the presence of the much larger dairy source. The PMF analysis, though, clearly establishes an evaporative and fugitive source that contributes to emissions of lighter hydrocarbons. This supports the conclusion that the majority of the CH₄ is being separated at the point of extraction from the associated gas and is not released with fugitive emissions (Gentner et al., 2014a). Kern County produces 75 % of all the oil produced in California (~ 6 % of US production) and has 81 % of the state's 60 000+ active oil wells (CDC, 2013). There is, however, a surprising scarcity of measured data to quantify the estimates of fugitive CH₄ from the prolific oil fields in the county and validate the bottom-up, activity-data-based inventory. Currently, fugitive emissions from fossil fuel extraction and distribution contribute ~ 5 % to the county's CH₄ emission inventory (KernGHG, 2012). Nationwide, a number of recent studies have reported significantly higher emissions of fugitive CH₄ from oil and gas production operations in other regions (Pétron et al., 2012; Karion et al., 2013; Miller et al., 2013; Kort et al., 2014). The PMF apportionment in this study (~ 2 %) is consistent with the fraction of fugitive CH₄ emissions in the regional and state inventories, but the PMF method, by itself, cannot accurately constrain emissions from minor sources owing to the uncertainties in the dominant sources.

We find that the vehicle emission source factor identified in this study makes no detectable contribution to observed N₂O enhancements. Our findings do not agree with the significant contribution (~ 18 %) of the transportation sector to the state's N₂O emission inventory (CARB, 2013). Vehicle dynamometer studies have indicated rapidly declining N₂O EFs with advancement in catalyst technologies, declining sulfur content in fuel and newer-technology vehicles (Huai et al., 2004). N₂O emissions from California vehicles, required to meet progressively stringent emission standards, are expected to decline and should have a minimal contribution to the CARB inventory in this decade. However, it seems the updates to the mobile N₂O emission inventory is not keeping pace with the improvements in vehicle catalyst technologies and corresponding decline in tailpipe N₂O emissions. Bakersfield is an urban region with a fairly large population (~ 500 000), and the essentially non-existent contribution of the PMF vehicle emission source to the N₂O apportionment and large divergence of the PMF-derived N₂O/CO EF from the state inventory EF for motor vehicles is a significant outcome pointing to overestimation of N₂O from motor vehicles in the inventory.

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